

Fine structure:

The H atom spectrum showed "fine structure"
In particular H_{α} showed doublet structure

Perturbations:

- o relativistic
- o spin-orbit
- o Darwin term

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$$H = T + V = \frac{p^2}{2m_e} - \frac{e^2}{r}$$

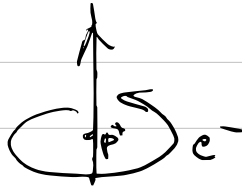
but relativistic is

$$T = \sqrt{p^2 c^2 + m_e^2 c^4} - m_e c^2$$
$$= \frac{p^2}{2m_e} - \frac{p^4}{8m_e^3 c^2} + \dots$$

• spin-orbit

In the frame of the electron the revolving proton produces a magnetic field.
The electron spin ~~it~~ interacts with it

Magnetic moment of orbit.



Magnetic moment of electronic orbit

$$M_L = \frac{i \pi r^2}{c} \quad i = \frac{e v}{2 \pi r} \quad \text{current}$$

$$= \frac{1}{2} \frac{e v r}{c}$$

$$= \frac{e}{2 m_e c} m_e v r \quad L = m_e v r$$

$$M_L = \frac{e}{2 m_e c} L$$

For historical reasons:

negative because of e^-

$$\vec{M}_L = - \frac{e \hbar}{2 m_e c} g_L \frac{\vec{L}}{\hbar}$$

In this case $g_L = \text{orbital } g \text{ factor} = 1$

$$\mu_B = \frac{e \hbar}{2 m_e c} = \text{Bohr magneton}$$

In a similar fashion

$$\vec{M}_s = -g_s \mu_B \frac{\vec{S}}{\hbar}$$

where \vec{S} = spin
angular momentum

$$\frac{\vec{S}}{\hbar} = \pm \frac{1}{2}$$

but $g_s = 2.002318$

In the frame of the electron

$$\vec{B} = \frac{(-Ze\vec{v}) \times \vec{r}}{cr^3}$$

The electron is acted upon by the electrostatic force

$$\vec{E} = \frac{Ze}{r^3} \vec{r}$$

$$\therefore \vec{B} = -\frac{L}{c} \frac{\vec{v} \times \vec{E}}{c}$$

but

$$\vec{E} = \frac{-\nabla\phi}{e} = \frac{1}{e} \frac{dV}{dr} \frac{\vec{r}}{r} \quad -\nabla\phi = \text{force}$$

Thus $\vec{B} = \frac{1}{emec} \frac{dV}{dr} \frac{1}{r} \vec{L} \quad L = m_e \vec{v} \times \vec{r}$

The interaction with electron is

$$E = -\left(\frac{L}{2}\right) \vec{M}_S \cdot \vec{B} \quad \rightarrow \text{Thomas Precession}$$
$$= \frac{1}{2m_e c} \frac{1}{r} \frac{dv}{dr} \vec{S} \cdot \vec{L}$$

Note $\frac{dv}{dr} \propto \frac{1}{r^3}$

$$\left\langle \frac{1}{r^3} \right\rangle = \frac{Z^2}{n^3 a_0^3} \frac{1}{l(l+\frac{1}{2})(l+1)}$$

$$\langle \vec{L} \cdot \vec{S} \rangle = \frac{\hbar^2}{2} [j(j+1) - l(l+1) - s(s+1)]$$

$$\langle H_{so} \rangle = \frac{E_n^2}{m_e c^2} n \left[\frac{j(j+1) - l(l+1) - 3/4}{l(l+\frac{1}{2})(l+1)} \right]$$

For relativistic correction

$$\langle H_r \rangle = \frac{-E_n^2}{2m_e c^2} \left(\frac{4n}{l+\frac{1}{2}} - 3 \right)$$

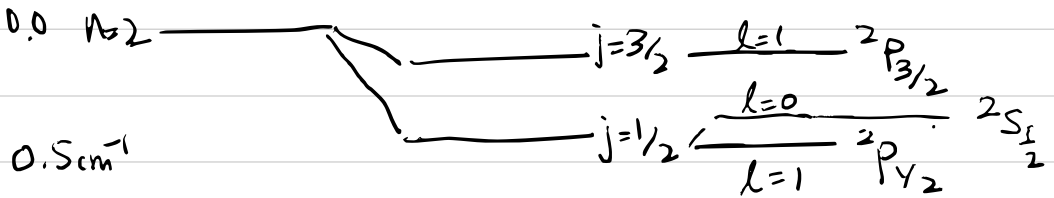
Add the Darwin correction to find

$$E_{nj} = E_n \left[1 + \left(\frac{Z\alpha}{n} \right)^2 \left(\frac{n}{j + \frac{1}{2}} - \frac{3}{4} \right) \right]$$

Dirac Theory:

note $(Z\alpha)^2$

$$E_{jn} = -m_e c^2 \left[1 - \left(1 + \frac{\alpha^2}{\left(n - j + \frac{1}{2} + \sqrt{\left(j + \frac{1}{2} \right)^2 - \alpha^2} \right)^2} \right)^{-1/2} \right]$$



Lamb



SPECTROSCOPIC NOTATION

$$2S+1 \boxed{L}$$

$$J$$

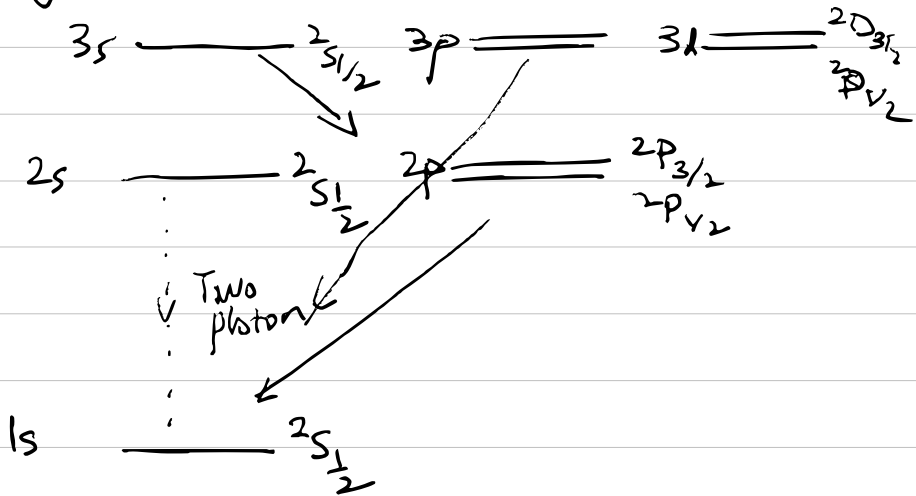
$$L = 0, 1, 2, 3$$

$$l = 0, 1, 2, \dots$$

$$\boxed{L} = S \ P \ D \ F$$

$$\boxed{l} = s, p, \dots$$

Hydrogen:



Allowed for forbidden
Two photon transition

$$\Delta l = \pm 1$$

$$\Delta j = 0, \pm 1$$

$$\Delta n > 0$$